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Pg. 7.4 $Re = \frac{\rho_{air} dv}{\mu} = \frac{1.29 \times 10^3 \text{ g/m}^3 \times 20 \times 10^{-6} \text{ m} \times 0.0127 \text{ m/s}}{0.0172 \text{ m/s}}$
 $= 0.02$ So for all of these particles, the

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Reynolds number is much less than 1, which means (7.24) is a reasonable approximation to the settling velocity. 7.14 Finding the percentage by weight of oxygen and the fraction (by weight) of oxygenate needed to provide 2% oxygen to the resulting blend of gasoline.

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